S/080/61/034/002/008/025

Effect of ultrasonic waves ...

The present authors point out that the mechanism of the effect of ultrasonic waves on electroplating, especially of alloys, is of interest for further investigations. There are 11 figures, 3 tables and 12 references: 7 Soviet-bloc and 5 non-Soviet-bloc. The two English-language publications read as follows: Fishlock, Metal Industry, 93, 109 (1958), St. R. Rich, Plating, 42, 11 (1955).

SUBMITTED: June 18, 1960

ACCESSION NR: AP4032501

s/0080/64/037/004/0800/0806

PARTIES OF THE PROPERTY OF THE

AUTHORS: Fel'dman, Yu. A.; Shatsova, S. A.; Gudkova, Ye. Ye.

TITLE: Nickel plating under the action of an ultrasonic field

SOURCE: Zhurnal prikladnoy khimii, v. 37, no. 4, 1964, 800-806

TOPIC TAGS: nickel plating, electroplating, ultrasonication, cathodic polarization, electrodeposit porosity, electrodeposit adhesion, current yield

ABSTRACT: The effect of ultrasonics on nickel plating from concentrated nickel sulfate solutions was examined. Experiments were run plating nickel from solutions containing 200-250 and 500 gm/l NiSO, plating nickel from solutions containing 200-250 and 500 gm/l NiSO, but of a frequency of 15-16 kilocycles/sec. It was found the field of a frequency of 15-16 kilocycles/sec. It was found the electrolytes containing 250 or 500 gm/l NiSO, 7H2O were not stable and needed constant correction of pH; their current yield was lower and needed constant correction of pH; their current yield was lower (75-85%), and the more concentrated electrolyte could not be sonicated when its depth was more than 10 cm. The electrolytes containing 200-250 gm/l NiSO, 7H2O, 30 H5BO3, 10 NaCl, 4NaF (and possible

Card 1/2

ACCESSION NR: AP4032501

formalin and naphthalene disulfonic acid) gave current yields of 96-98% under sonication. The maximum permissible current density was increased three times (at 20C) to five times (at 50C) by sonication. Cathodic polarization was also reduced somewhat. Use of ultrasonics during the electroplating does not affect the adherence of the plate to the base metal, but does reduce the porosity of the deposit. "M. V. Kurganova and A.K. Mokshantseva took part in conducting the experimental work." Orig. art. has: 3 figures and 2 tables.

ASSOCIATION: None

SUBMITTED: 30Dec62

ENCL: 00

SUB CODE: MM

NR REF SOV: 013

OTHER: 006

Card 2/2

20709 112 /61/00 /001/050/062 Tpatkin, 1 S., Stepanov B.M., and Shatsukevi h. A F. 9,4130 (1138, 1141, 2801, 3201) Photomuliabler Detection of Geray Pulses PENCONICAL Pribory & tekhnika eksperimenta, 1961 No. 1 pp 165-166 A large number of papers have been published in recent AUTHORS: years giving descriptions of the Narray unless the array tuber years Styling descriptions of warrous purson consents for a tubes. The form of the X-ray pulse is deadly recorded by a mint constitution from the X-ray pulse is deadly recorded by a photomultiplier feeding an amplifier and a fast oscillograph.
The research section of the Assay in the description of the feeding and applifier and a fast oscillograph. photomultiplier recurs an amplifier and a transfer the ferm of the use of an amplifier introduces a distortion lute the ferm the the recorded the recorded X-ray pulse and complicates the measurements. present authors report preliminary results of a study of the ferm present authors report programmary results of a study of the of X-ray pulses obtained without the use of an amplifier, and the of the fort of the order of the or nrw-8 (PGI-8) electron multiplier and the cy 19 (OK-19M) oscillograph were employed. The form of X-ray pulses from a soutimental moment demonstrate who was investigated continuously pumped demountable Repay tube was investigated. The tube voltage was derived from the Total or portage consists of formal table and the continuously pumped derived from the continuously pumped derived from the continuously pumped. The tube voltage was derived from the (Abendo) (ULA-200) pursed-voltage generators. The election multiplier PGI-8 consists of four voltage generators with remonstrations are as in each vortage generators with ten multiplying stages in each channels with ten multiplying stages in each channels with ten multiplying stages. emitters and cathodes were used. They have a quantum yield of card 1/4

20709

\$/120/61/00 /001/050/062 £032/E114

Photomultiplier Detection of X-ray Pulses

1.5 x  $10^{-3}$  for slev electrons and X-ray energies hereign 0.2 and 1.5 LeV. The amplification coefficient is 107/108 and the applied voltage 500 volts per stage. The multiplier emput is developed across a 75 observatial cable. The maximum output current per pulse is not less than 5 amps so that the signal can be applied directly to the oscillograph. The dependence of the form and duration of the X-ray pulses was investigated as a function of the material and form of the cathode the distance between the cathode and the anode and the pressure in the tube. The figure shows oscillegrous of X-ray pulses as functions of the distance between the electrodes for cathodes in the form of aluminium (1) and molyhdonum (2) needles, and a tantalum ring with a sharp rim (5). The anode of the tube was in the form of a plane molybdemus disc. The calibration trace on the photographs is a 10 Mc/s signal. The distance between the electrodes was varied between 55 mm (upper photographs) and 5 mm. As can be seen, the duration of the K-ray pulse decreases as the electrodes approach each other. The form, duration and amplitude of the Card 2/4

20709

S/120/61/000/001/050/062 E032/E114

Photomultiplier Detection of X-ray Pulses

pulses was found to be independent of the cathode material. the tube incorporates a nitrogen trap, the form of the pulse remains stable when the pressure is increased to 10-3 mm Hg. the tube is operated without the trap, the stability deteriorates. The optimum working conditions of the tube at a working voltage of 470 kV per pulse were: pressure 10-5 mm Hg, anode to cathode distance 25 mm. The amplitude of the pulse under these conditions does not vary by more than ± 3% over long periods of time. The total output of X-rays is then 1019 - 1020 quanta/sec with a pulse There are 1 figure and 5 references: 2 Soviet and 3 non-Soviet.

A THE SHARE SHOWER SHOW

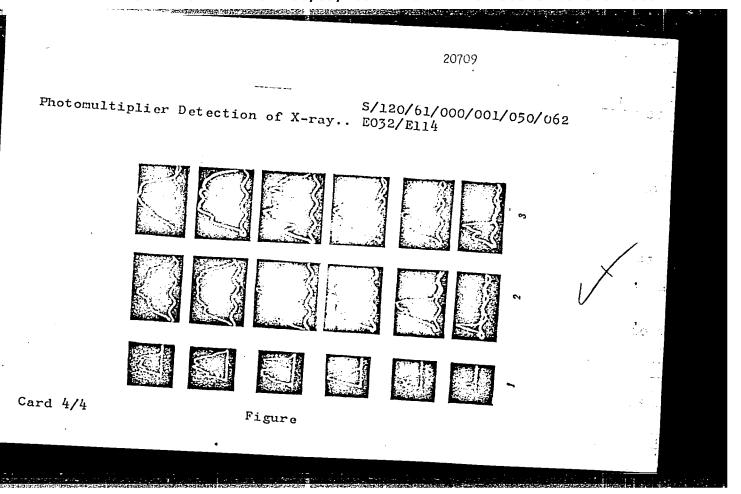
Institut khimicheskoy fiziki AN SSSR

(Institute of Chemical Physics, AS USSR) SUBMITTED:

June 24 1959, and in final form December 19, 1959

Card 3/4

10



LUTHAR ! V:107-5:-10-39/65 a lorgable Joshine! Lastmusent (Portatively Normalingvanny) TIALS: nii or PE:I DIUMb: | Audio, 1950, Nr 10, 25 40-49 (2004) At the Nit all-Union Exhibition of the Greative Forh of ama-teur Labor 1 ignors of the POSIMF in Righ, the author showed ABSTRACT: a billistary be tabler for neasuring the frequency of a cirsuit; it can what he are an elemeter and voltage or, a merometer and meaning receivers.
The design of the nultimeter is a conditation of a TIR covering a frequency band of 4.0 km - 51 cm, a volucator for do and me, papable of measuring currents from 0 to 15 v, 6 to 150 v and f to 600 v, and an obmmater for measuring resistances from to o'me to 2 mesorms. A full description is given-There are 2 findres and 1 directit dis ram. AND CHANIST Stallinging o'd entropy of inches (Stalin) (Sheet Radio Nat) Card 1,1

SHATUKH, V.G.

Role of association in the process of memorizing a vocabulary of foreign words. Nauk. zap. Nauk.-dosl. inst. psykhol. 11:115-119 '59. (MIRA 13:11)

1. Gosudarstvennyy universitet im. I. I. Mechnikova, Odessa.
(Association of ideas)
(English language—Study and teaching)

DIKSHTEYN, Ye.I., MAGIESON, M.A., SHATUKHOV, A.I., GAZHUR, V.F.

Improving the luminance and organizing the natural gas fuel spray. Stal\* 24 no.10:890-892 0 \*64. (MIRA 17.12)

1. Magnitogorskiy metallurgicheskiy kombinat i Chelyabinskiy nauchno-issledovatel'skiy institut metallurgii.

Water, a

USSR/Inorganic Chemistry. Complex Compounds.

C

Abs Jour : Ref Zhur - Khimiya, No. 8, 1957, 26441.

Author Vol'nov, I.I., Shatunina, A.N. Inst Academy of Sciences of USSR. Title Possibility of Preparing Ba(02)2.

Orig Pub : Dokl. AN SSSR, 1956, 110, No. 1, 87 - 38.

Abstract Assuming that the structure of  $Ba(0_2)_2$  was similar to the structure of superperoxides of alkali metals, the authors estimated the value of the equilibrium constant of the value of the equilibrium constant of the reaction  $BaO_2$  (solid)  $O_2$  (gas)  $\rightleftharpoons Ba(O_2)_2$  (solid) (1). Using the approximately determined values of  $\triangle H298 = -14.2$  kcal and of  $\triangle S298 = -45.4$  entr. units, the equilibrium pressures of  $O_2$  were computed for the reaction (1) at 25, 100, 200, 210, 220, 230, 250 and  $302^0$ , which were 32, 75, 2300, 3150,

Card 1/2

USSR/Inorganic Chemistry. Complex Compounds.

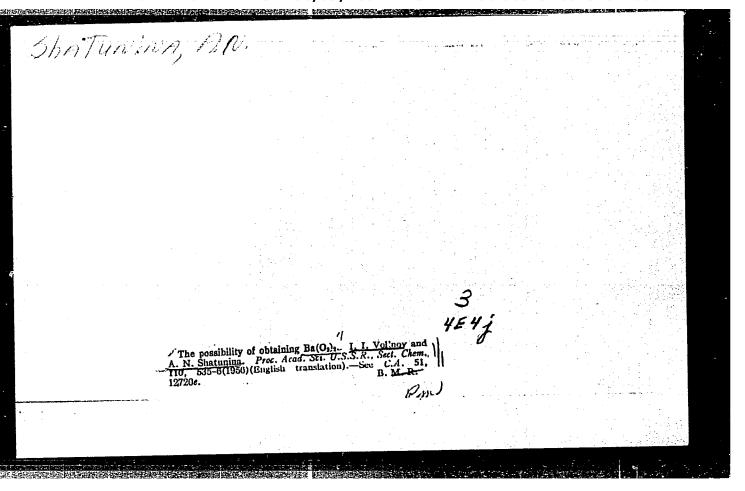
NEW TOTAL SERVICE AND BUILDING TO THE PROPERTY OF THE PROPERTY

C

Abs Jour : Ref Zhur - Khimiya, No. 8, 1957, 26441.

4230, 7278, 9760 and 33600 atm. This allows to expect that the reaction (1) is possible at superhigh pressures and a high temperature. It is shown that up to 10.4% by weight of Ba(0<sub>2</sub>)<sub>2</sub> are forming in Ba0<sub>2</sub> exposed to the pressure of 0<sub>2</sub> at 3300 atm at 210° for the duration of 19 hours. As it seems, the reaction takes place on the boundary between the solid and the gaseous phases.

Card 2/2



VOL'NOV, I.I.; SHATUNINA, A.N.

Reactivity of superoxides of alkaline earth metals with H<sub>2</sub>O and CO<sub>2</sub>. Zhur.neorg.khim. 2 no.7:1474-1478 J1 '57. (MIRA 10:11)

1. Institut obshchey i neorganicheskoy khimii im. N.S.Kurnakova AN SSSR.

(Alkaline earths) (Water) (Carbon dioxide)

VOL'NOV, I.I.; SHATUNINA, A.N.

Formation of LiO<sub>2</sub> from Li<sub>2</sub>O<sub>2</sub>·2H<sub>2</sub>O<sub>2</sub>. Izv. AN SSSR Otd. khim. nauk no.6:762-763 Je <sup>7</sup>57. (MIRA 10:11)

1. Institut obshchey i neorganicheskoy khimii im. N.S. Kurnakova AN SSSR.

(Lithium oxides)

AUTHORS: Vol'nov I. I. Chamova, v. I. Chat wing, A. A. 78-3-5-6/39 TITLE: V.The Permeties of  $J_{3}(0_{2})_{2}$  by Irradiation of  $CaO_{2}$ .  $2H_{2}O_{2}$  With Ultra-Violet Rays (V. Obrazovaniye  $Ca(0_{2})_{2}^{2}$ oblinsheriyen 0202.22202 ultraficletovymi luchami) PERIODICAL: Ehurnal Neorganisheskoy Khimii 1958, Vol. 3. Nr. 5, FF 1095-1097 (USSR) ABSTRACT: By irradiation of CaO2.2H2O2withultra-violet light,  $Ca(O_2)_2$  is obtained the presence of which can only be verified by chemical analysis. The obtained preparation is lemon-coloured, similar to NaO2. The final product is not uniform. The manufacturing process is difficult to reproduce. In some cases the final product contains from 7-8%  $Ca(0_2)_2$ , and in others it has a higher content of from 10-14/3.  $Ca(0_2)_2$  was also obtained by irradiation of  $Ca0_2.2H_20_2$ in a Fischer pistol consisting of quartz-glass, at a Card 1/2 temperature of 25-450c.

The Formation of  $Ca(\theta_2)_0$  by Irradiation of  $Ca\theta_2.2H_2\theta_2$  78-3-5-5/39

 $\operatorname{Ca}(\mathbb{Q}_2)_2$  is very unstable. There are ! table and 0 references, 5 of which are

AGJOCIATION:

Institut obshehey i neurganicheskoy khimii im. N. 3. Kurnakova Akademii nauk BBSR (Institute of General and Improved Chemilatry imeni L. S. Kurnakov, AS USSR)

SUBMITTED: May 10. 1957

AVAILABLE: Library of Congress

> 1. Calcium chiles. Treduction 2. Calcium chides. Properties 3. Ultraviolet regentifications

Card 2/2

### CIA-RDP86-00513R001548720007-7 "APPROVED FOR RELEASE: 08/09/2001

sov/78-4-2-3/40

Volinov, I. I., Shatunina, A. N. 5(2) AUTHORS:

The Formation of Lithium Superperoxide From Li<sub>2</sub>0<sub>2</sub>.2H<sub>2</sub>0<sub>2</sub>

(Obrazovaniye nadperekisi litiya iz Li<sub>2</sub>0<sub>2</sub>.2H<sub>2</sub>0<sub>2</sub>) TITLE:

Zhurnal neorganicheskoy khimii, 1959, Vol 4, Nr 2, PERIODICAL:

pp 257-259 (USSR)

Lithium peroxide is formed on drying Li202.2H202 in vacuo ABSTRACT:

at 100-120° and 10 Hz . A thin layer of the  $\text{Li}_2\text{O}_2\text{-}2\text{H}_2\text{O}_2$ sample was laid on the crystallizer surface and was dried. The change of the percentage of the LiO2 content of the

preparation in dependence on temperatures of 100°, 120°, and  $70^{\circ}\text{C}$  and time, was investigated. The results are shown in figure 1. There is 1 figure, 1 table, and 12 references,

8 of which are Soviet.

Institut obshchey i neorganicheskoy khimii im. N. S. Kur-ASSOCIATION:

nakova Akademii nauk SSSR (Institute of General and Inorganic Chemistry imeni N. S. Kurnakov of the Academy of

Sciences, USSR) Card 1/2

5(2)

SOV/78-4-7-5/44

AUTHORS:

Vol'nov, I. I., Shatunina, A. N.

TITLE:

The Forming of NaO $_2$  From Na $_2$ O $_2$ ° 2H $_2$ O $_2$  (Obrazovaniye NaO $_2$  iz

Na<sub>2</sub>O<sub>2</sub>•2H<sub>2</sub>O<sub>2</sub>)

PERIODICAL:

Zhurnal neorganicheskoy khimii, 1959, Vol 4, Nr 7,

pp 1491-1493 (USSR)

ABSTRACT:

The compound Na202.2H202 was produced according to the slightly

altered method by F. D'Ans (Ref 3) by the action of 98-99% hydrogen peroxide upon solid CH3Na.O.5.CH3. After the decay

temperature of the crystalline compound had been determined as amounting to 1520 by plotting a heating curve of this compound (Fig 1), the compound was dried in a vacuum at 70, 100 and 1200, i.e. below decay temperature. The peroxide-oxygen was determined gasometrically according to E. Seyb (Ref 4). The experimental results are given by table 1 and figure 2. Preparations with a content of about 30 weight % NaO, were obtained, which is con-

siderably more than has hitherto been known in publications.

Card 1/2

There are 2 figures, 1 table, and 4 references.

sov/78-4-7-5/44

The Forming of NaO2 From Na2O2.2H2O2

ASSOCIATION:

Institut obshchey i neorganicheskoy khimii im. N. S. Kurnakova Akademii nauk SSSR (Institute for General and Inorganic Chemistry imeni N. S. Kurnakov of the Academy of Sciences,

USSR)

April 4, 1958 SUBMITTED:

Card 2/2

#### CIA-RDP86-00513R001548720007-7 "APPROVED FOR RELEASE: 08/09/2001

5/052/63/000/002/001/020 B144/B186

AUTHORS :

Wol'nov, 1. I., and Shabunina, A. S.

TITLE

By drates of endium notabilicate hydroperoxidates

PERIODICAL: Abademiya nauk SSSE. Izvestiya. Otdeleniye khimichezkikh

nauk, no. 2, 1963, 201 - 205

TEXT: Phase dlagrams were plotted for the MagSiOg-HgO system at C and 1500 to determine the exact composition of the hydrates of sedium meta-silicate hydropercyldates. The initial substances sere Re2SiO, 9H2O dohydrated to Na<sub>2</sub>SiO<sub>3</sub>·6H<sub>2</sub>O and dilute H<sub>2</sub>O<sub>2</sub>. To avoid vigorous H<sub>2</sub>O<sub>2</sub> reactions, the "residues" obtained with week R202 solutions at 0°C were used for the high  $H_2O_2$  concentration ranges. Equilibrium was obtained at 6 - 10 hrs. The liquid and solid phases were enalyzed quantitatively for active 02: Na<sub>2</sub>O, and SiO<sub>2</sub>. The solubility isotherm at o<sup>O</sup>C proved the existence of 3 eolld phases: NagSiO3 9H2C with 0 - 4 % by weight H2O2, Re2SiO3 2.5H2O2H2O card 1/2

Hydrates of sodium metasiliceta...

E/060/63/000/002/001/020 B144/B186

with 4 - 20 % by weight  $H_2O_2$ , and  $H_{S_2}S_1O_3 \cdot 3H_2O_2 \cdot H_2O$  with  $20 - 34 \lesssim b_2$ weight H202. For the investigation at 15°C, Na23103.3H202.H20 was taken as initial substance. Equilibrium was obtained after 4 - 5 hrs. The phase diagram shows 2 solid phases: Ne 2510, 9H202 H20 with 0 - 5 % by weight  $\rm H_2O_2$  and  $\rm Ma_2SiO_3 \cdot 5H_2O_2$  with 5 - 42 % by weight  $\rm H_2O_2$ . The heating curve plotted for Neg5103-3H252' H20 using a Surnakov differential pyrometer showed an exchangio effect at  $\sim 60^{9}$ C inclusions the decomposition of the crystallized  $\rm H_2O_2$ , and an andothermic effect due to dehydration at  $\sim 120^{9}$ C. The general formula established for these sempounds to Na SiO3 - xH2O2 - yH2O1 where x = 2.5 = 3, and y = 0 = 1. There are 4 figures and 2 tables.

ASSOCIATION: Institut obshehoy i necessanisheekoy kuimii im, R. S. Eurmakova Akademii nauk SSOR (Inclitate of General and Inorganic Chemistry imant N. F. Furnakov of the Accommy of Sciences USSR)

SUBMITTED: Day 18, 1962

Card 2/2

ACC NR: AP7000813

SOURCE CODE: UR/0062/66/000/011/2032/2033

AUTHOR: Vol'nov, I. I.; Shatunina, A. I.

ORG: Institute of General and Inorganic Chemistry im. N. S. Kurnakov, Academy of Sciences, SSSR (Institut obshchey i neorganicheskoy khimii Akademii nauk SSSR)

TITLE: Now data on the formation of  $Ca(O_2)_2$  via  $CaO_2 \cdot 2H_2O_2$ 

SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 11, 1966, 2032-2033

TOPIC TAGS: calcium compound, superoxide

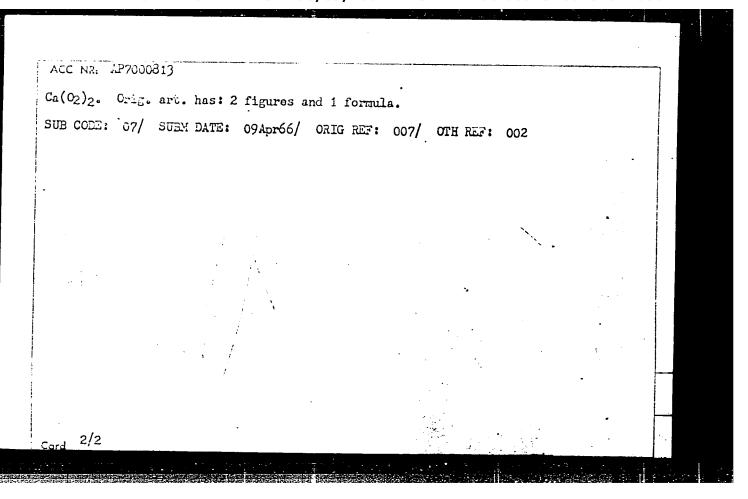
ABSTRACT: The reaction

 $2C_aO_2 \cdot 2H_2O_2 \rightarrow C_a(O_2)_2 + C_a(OH)_2 + 3H_2O + 1.5O_2$ 

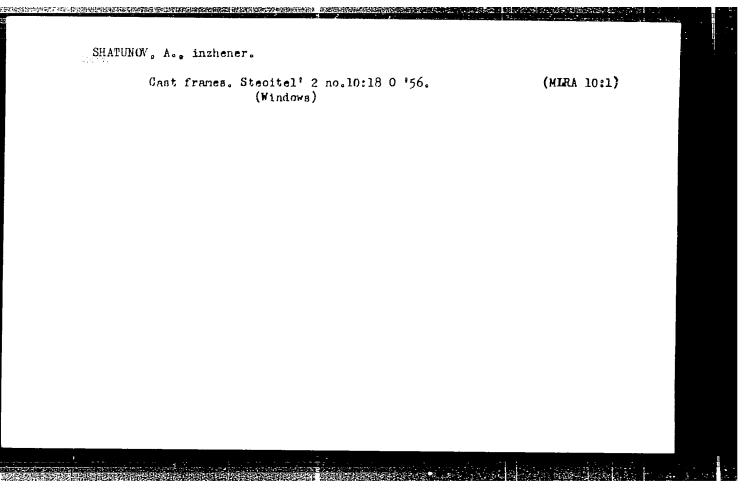
was carried out at reduced pressures, a constant temperature of 40 °C, a constant time of 60 min, and with a distribution of CaO2.2H2O2 of 1 g over 100 cm2 of surface. It was found that the maximum content of  $Ca(O_2)_2$  in the end product, equal to 55.4 wt. was found that the maximum content of  $Ca(O_2)_2$  in the end product, equal to 55.4 wt. %, is reached at a pressure of 6 x 10-3 mm. The dependence of the  $Ca(O_2)_2$  content on the surface over which 1 g of  $CaO_2 \cdot 2H_2O_2$  was spread at a residual pressure of the surface over which 1 g of  $CaO_2 \cdot 2H_2O_2$  was spread at a residual pressure of  $CaO_2 \cdot 2H_2O_3$  was determined. The value obtained, 55.4 wt. %  $Ca(O_2)_2$  at 100 cm<sup>2</sup> of surface, is close to the limiting value, since according to the reaction given above. 1 face, is close to the limiting value, since according to the reaction given above, 1 mole of Ca(C2)2 and 1 mole of Ca(OH)2 are formed, which corresponds to 58.4 wt. %

Card 1/2

UDC: 661.842.24:542.91



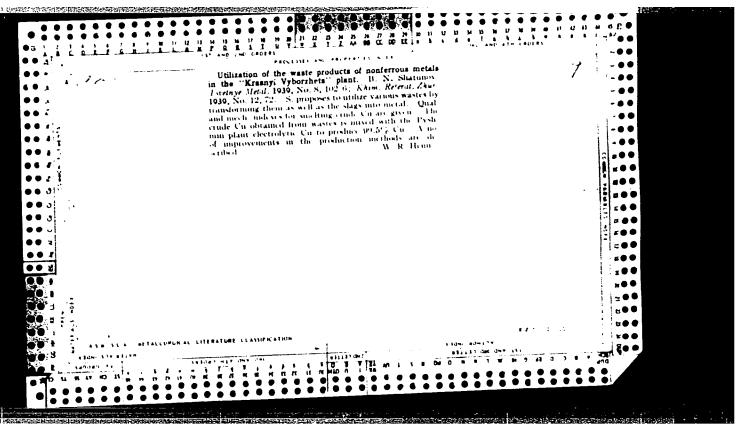
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9.	Mo	nthly	List	<u>of</u>	Russian	Acces	sions,	Library	of	Congre	ss, _	May	 1953.	Unclas	sified.	
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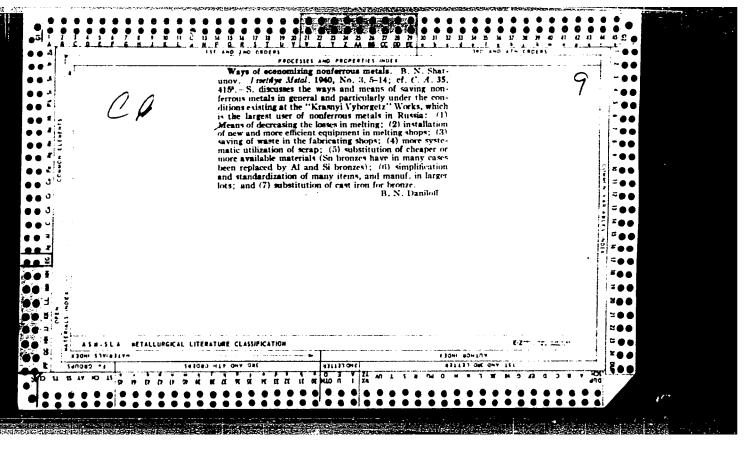


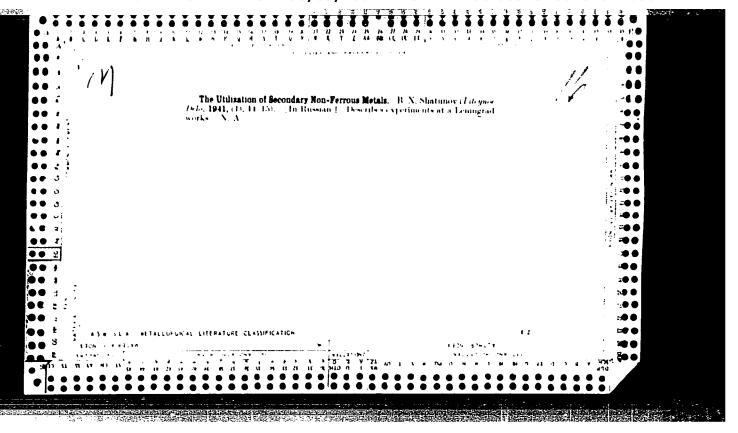
SHATUNOV, B.

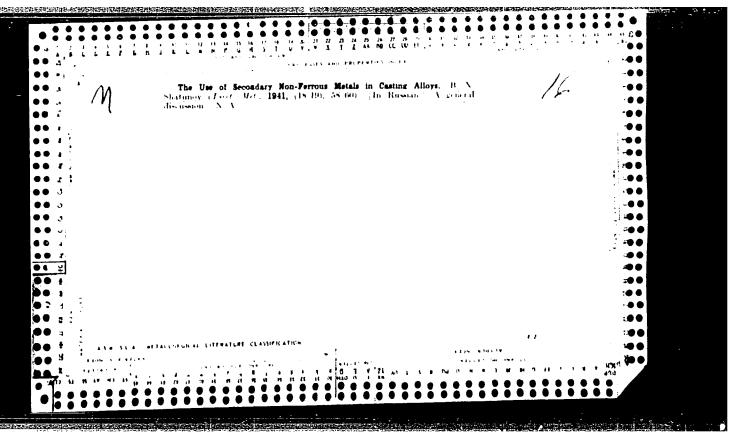
Contribution of Leningrad metallurgy to exports. Vnesh. torg.
41 no.8:33 '61.

1. Zamestitel' nachal'nika otdela upravleniya metallurgicheskoy promyshlennosti Lensovnarkhoza.
(Russia—Commerce)
(Leningrad—Metal industrics)









SHATUNOV, Boris Nikolayevich; KORNYUSHIN, M.Ya., inzhener, retsenzent; SHAPOSHNIKOV, V.A., inzhener, retsenzent; RL'DIND, L.M., redaktor izdatel'stva; KVENSON, I.M., tekhnicheskiy redaktor

[The manufacture of aluminum ware] Proizvodstvo aliuminievoi posudy.

Moskva, Gos. nauchno-tekhn. izd-vo lit-ry po chernoi i tsvetnoi
metallurgii, 1956. 176 p.

(Kitchen utensils) (Aluminum)

S/135/60/000/009/014/015 A006/A002

AUTHOR:

Shatunov, B. N. Engineer

TITLE:

Butt Welding of  $6p.0 \oplus 6.5-0.15$  (Br.OF 6.5-0.15) Bronze Strips for

Cold Rolling

PERIODICAL. Svarconnoye proizvodstvo, 1960, No. 9, pp. 40-41

During the past years the rolling shops have been equipped with fully mechanized high-speed cold rolling mills. Rolling of light-weight blanks on such mills proved to be economically disadvantageous. In this connection welding of the blanks prior to cold rolling is highly important. Butt welding of several 3 mm thick 3 m long strips into one blank was employed at the "Krasnyy Vyborzhets" Plant to extend the continuity of the high-speed cold rolling process. Presently, 7 to 8 strips are welded into a blank of 20 - 25 m total length. The amount of welded strips may be raised. Continuous welding is performed as follows: the strips are prepared at the etching department, cleaned and cut with guillotine shears. Welding is performed on a movable table where the strip butts are joined and clamped. After welding the blank is turned into a drum with the aid of a reductor from which the blank is removed

Card 1/2

3/135/60/000/009/014/015 A006/A002

Butt Welding of  $\overline{b_p}$ ,  $0\phi_{6.5}$ -0.15 (Br OF 6.5-0.15) Bronze Strips for Cold Rolling

by a pneumatic device. Operation of the shears, switching on the welding machine, the reductor and the remover is controlled from a desk placed besides the welding operator. Buth welding is carried out by argon-arc process with tungsten electrode. The arc is fed by d-c of reversed polarity. The welding head is water-cooled. The welding conditions are: 180 - 200 amps, 18 v arc voltage, 8 1/min argon consumption, 3 8 mm electrode diameter and 36 m/nr welding speed. The method yielded a raise in the production from 50 to 66.9% and yearly savings of 112,700 rubles.

Card 2/2

Mechanization of clay-mining operations. Mekh.i avtom.proizv.

14 no.8:39-40 Ag '60.

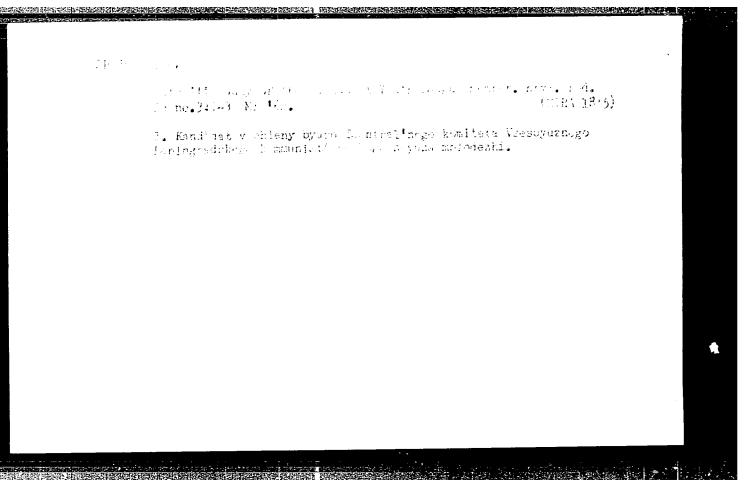
(Clay) (Mining machinery)

(Clay)

зватилот, G.

Train the loyal sond of the fatherland. Kryl. rod. 16 no.11:6-7 N '65.

1. Chlen Prezidiura TSentral nogo komiteta Vsesoyuznogo dobrevol nogo obshchestva sodeystviya armii, aviatsii i flotu SSSR.



Constant attention to the military and patriotic education of our youth. Voen.com. 30 no. 10:3-5 0 %3. (MIRA 16:11)

1. Chien prezimina monatral and komiteta kobrovol acgo coshrbent va sodeystviya armii, aviatsil i flotu.

SHATUNOV, G., general-mayor

Jointly with the people and the army. Radio no.4:8-9 Ap 165.

(MIRA 18:5)

SHATUROV, G.

The Spartekiada calls for new frontiers. Eryl. rod. 16 no.1:
1.3 Ja '65. (MIRA 12:3)

1. Chlen Prozidiuma TSentral'nogo komiteta Vsesoyuznogo
dobrovol'nogo obshchestva sodeystviya armin, aviatsii i flotu.

SHATUNOV, G.

Combat missions of the All-Union Volunteer Society for rural areas. Assistance to the Army, Navy, and Fir Force in rural areas. Voen. znan. 41 no.813-4 Ag '65. (MIRA 1817)

DESCRIPTION OF THE PROPERTY OF

1. Chlen prezidiuma TSentral'nogo komiteta Vsesoyuznogo dobrovol'nogo obshchestva sodeystviya armii, aviatsii i flotu.

SHATUNCY, G. F.

Komsomol'skie organizatsii sovetskoi Armii - peredovoi otriad armeiskoi molodezhi Communist Youth League organizations of the Soviet Argy are the advance detail of the army youth. Moskva, Voennoe Izd-vo, 1953. 132.

SO: Monthly List of Russian Accessions, Vol. 6, No. 5, August 1953.

#### CIA-RDP86-00513R001548720007-7 "APPROVED FOR RELEASE: 08/09/2001

Shatones, 6.

AUTHOR:

107-8-2/62 Shatunov, G., Vice-President of TsK-DUSAAP

TITLE:

General Development of Socialist Competition among Radio Amateurs (Vsemerno razvivat' sotsialisticheskoye sorevnovaniye

eredi radiolyubiteley)

PERIODICAL:

Radio, 1957, #8, pp 1-2 (USSR)

ABSTRACT:

The radio amateurs play an important role in the DOSAAF. They encourage the growth and activity of the competitive radio amateur organizations and the number of radio clubs in enterprises, collective farms and schools is growing. Hundreds of thousands young men participate in this attractive and useful activity and acquire the necessary minimum knowledge of radio

For example, the members of the Omek radio club decided to prepare during 1957 50% more radio specialists than previously projected; to establish in Omsk and other regional centers 20 ultra-short-wave and 3 short-wave radio stations etc.

The experience of the best radio clubs and individual radio amateurs acquired during the competitions should be spread

Card 1/2

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107-8-2/62

TITLE:

General Development of Socialist Competition among Radio Amateurs (Vsemerno razvivat' sotsialisticheskoye sorevnovaniye sredi radiolyubiteley)

among all organizations. The magazine "Radio" and the newspaper "Sovetskiy Patriot" play a large role in this activity.

A recent resolution of the VIIth Plenum of the "TskDOSAAF" says that in numerous organizations, the collaboration of active members is still unsatisfactory and requires that all the committees, clubs and organizations improve their cooperation with active members, which is decisive for the further expansion of military science and the development of the educational and competitive activity.

In the near future, it is indispensable to increase the number of active radio amateur members as well as the number of public instructors.

One Russian reference is cited.

INSTITUTION: DOSAAF

PRESENTED BY: SUBMITTED:

AVAILABLE:

At the Library of Congress

Card 2/2

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#### CIA-RDP86-00513R001548720007-7 "APPROVED FOR RELEASE: 08/09/2001

5H17

85-9-3/33

AUTHOR:

Shatunov G., Deputy Chairman, Central Committee of the

DOSAAF, USSR

TITLE:

The Instruction Given to the Active Members of the DOSAAF

Organizations mustibe Radically Improved (Radikal'no

Uluchshit' Rabotu s Aktivom)

PERIODICAL: Kryl'ya Rodiny, 1957, Nr 9, pp. 2-3 (USSR)

ABSTRACT:

The author interprets the purport of the resolutions passed at the June 1957 Seventh Plenary Session of the Central Committee of the DOSAAF, which states that "the work of the DOSAAF organizations in the military field and in the field of sports has not yet reached the broad masses of the population to a degree required by the interests of our State, nor is the quality of that work high enough", and which requires that "measures be taken in order to improve the instruction given to the active members of the DOSAAF organizations". The author points

card 1/3

APPROVED FOR RELEASE: 08/09/2001 CIA-RDP86-00513R001548720007-7" The Instruction Given to the Active Members of the DOSAAF Organizations must be Radically Improved (Cont.)

了自然成功的现在分词的问题 医角膜 医克拉克氏试验检试验检验 法经历经济

out that the discussions which took place at this session have shown the necessity of supplementing the work of the staff members of the organizations and the work of the selected sportsmen-aviators by putting more stress upon the work of voluntary instructors - members of the DOSAAF recruited among the masses of the population. He asserts that it is therefore very important that "in conformity with the request of the Seventh Plenary Session the number of voluntary instructors be increased, and brought, in all branches of sports, up to the level of the needs of the DOSAAF organizations", and recommends that "DOSAAF Committees organize a systematic training of voluntary instructors, aimed at improving their special skills and at widening their knowledge in the field of methodology." He also advises that the training and education of these instructors be directed towards developing their initiative. The article discloses that it is planned to double, within the next three years, the number of the active members of the DOSAAF, and that these new members will be called to participate in the strengthening of the defenses of the

Card 2/3

The Instruction Given to the Active Members of the DOSAAF Organizations

country. The article contains no information of

scientific value.

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must be Radically Improved (Cont.)

Card 3/3

Active leadership is the wais for success in was defense work.

Voen.znen. 33 no.6:12-17 % ''''. (V.74 10:5)

1.Carestitel predictatelys Thenthallnogo Accidents Debrovelings obsochestva sodeyctyze armi, aviatsii i flotu SSS. ("ilitary education)

("ilitary education)

85-58-3-4/26

AUTHOR: Shatunov, G., Acting Chairman of the Central Committee of DOSAAF, USSR

TITLE: The 40th Anniversary of Lenin's Komsomol (K 40-letiyu Leninskogo komsomola); Strengthen and Develop Friendship Between the Komsomol and DOSAAF Organizations (Krepit' i razvivat' sodruzhestvo komsomola i DOSAAF)

PERIODICAL: Kryl'ya rodiny, 1958, Nr 3, pp 2-3 (USSR)

ABSTRACT: The author reviews the history of the Komsomol organization, established in 1922, and the role its membership has played in every phase of Soviet life and activity. He cites the record established by members of the VLKSM in various parts of the country and refers to Moscow, Irkutsk, and Gor'kiy oblasts and the Ukraine as areas where Komsomol members have participated most actively in DOSAAF. The author refers to the Spartacus games to be held between May and September [1958] to celebrate the 45th anxiversary of the VLKSM. Several million young sportsmen are expected to take part in these games, which will include

Card 1/2

3-2-1200

The 41th Anniversary of Lenin's Komsomol

85-58-3-4/26

competitions in marksmanship, automobile and motorcycle racing, parachote jumping, and in building airplane, auto and ship models. The author states that the XIII VLKSM Convention will be held on 15 April 1958 and that its decisions will stress the need for greater cooperation between Komsomol and DOSAAF organizations. Photo shows Komsomol member Nadezhda Borisova, glider pilot of the second Moscow Monacipal Aeroslub.

AVAILABLE: Library of Congress

Card 2/2

Consolidate and develop the results of the Spartakiada. Za rul.
16 no.10:2-3 0 58. (MIRA 12:1)

1. Zamestitel' predsedatelya TSentral'nogo komiteta Dobrovol'nogo obshchestva sodeystviya armii, aviatsii i flotu.

(Communist Youth League-Sports)

SHATUNOV, G., kand. v chleny byuro Tsentral'nogo komiteta Vsesoyuznego Leninskogo Kommunisticheskogo soyuza moledezhi.

The Communist Youth League and the defense of the country.

Voen. znan. 34 no.9:1-4 S \*58. (MIRA 11:10)

1. Zamestitel' predsedatelya TSentral'nogo Komiteta Vsesoyuznogo Dobrovol'nogo obshchestva sodevstviya armii, aviatsii i flotu SSSR. (Communist Youth League)

#### 

SHATUNOV, Georgiy Pavlovich; USPENSKIY, N.M., red.; FAYNSHMIDT, F.Ya., tekhn.red.

[All-Union Volunteer Society for Assistence to the Army, Air Force, and Navy of the U.S.S.R.] DOSAAF SSSR. Moskva, Izd-vo DOSAAF, 1959. 127 p. (MIRA 13:2)

(Military education)

SCV/107-59-4-5/45

AUTHOR:

Shatunov, G., Deputy Chairman

TITLE:

Expanding the Ranks of DOSAAF (Shirit' ryady DOSAAF)

PERIODICAL:

Radio, 1959, Nr 4, p 7 - 8 (USSR)

ABSTRACT:

The security of the USSR and the entire Soviet bloc demands that each Soviet citizen be constantly concerned with a further strengthening of the Soviet Armed Forces. Millions of Soviet citizens comply with this duty by participating in the activities of DOSAAF. The training of drivers and radio operators, conducted by DOSAAF, not only helps to increase the military preparedness of the USSR, but it simultaneously furnishes specialists for the national economy. However, DOSAAF is still far from its goal set at the 4th DOSAAF Congress, which calls for the participation of the majority of the adult population in this organization. The author points out that the Bryansk Oblast, DOSAAF Organization, headed by Chairman V. Nesmeyanov, is very

Card 1/2

SOV/107-59-4-6/45

Expanding the Ranks of DOSAAF

inactive in acquiring new members, since not even one new member joined this organization during a period of six months. Therefore the author demands that all members of the DOSAAF organizations intensify their efforts for increasing the number of DOSAAF members, whereby radio exhibitions and radio competitions may be used for attracting new people. There is 1 photograph. .

ASSOCIATION: Tsentral'nyy komitet DOSAAF SSSR (Central Committee

of DOSAAF USSR)

Card 2/2

APPROVED FOR RELEASE: 08/09/2001 CIA-RDP86-00513R001548720007-7"

SHATUNOV, G.

Radio engineering information should be introduced on a large scale to our youth. Radio no.2:3-4 F  $^{1}60$ .

(MIRA 13:5)

1. Kandidat v chleny byuro TSentral'nogo komiteta Vsescyuznogo Leninskogo kommunisticheskogo soyuza nolodezhi.
(Radio)

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Shartchov, G.

Spartskieda means mass participation and skill! Kryl.rod. 11
no.9:2-3 S '60. (MIRA 13:9)

1. éamestitel' predsedatelya Orgkomiteta Vsesoyuzaoy
spartakiady po tekhnicheskim vidam sporta.

(Sports)
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SHATUNOV, G.

Prepare to welcome the congress of the beloved party with suitable achievements. Za rul. 19 no.7:1-2 J1 '61. (MIRA 14:8)

1. Chlen prezidiuma TSentral'nogo komiteta Dobrovol'nogo obshchestva sodeystviya armii, aviatsii i flotu SSSR.

(Transportation, Automotive—Societies, etc.)

SHATUNOV, G.

Train young people in the spirit of preparedness for great deeds. Voenoznano 37 no.6:3-4 Je '61. (MIRA 14:6)

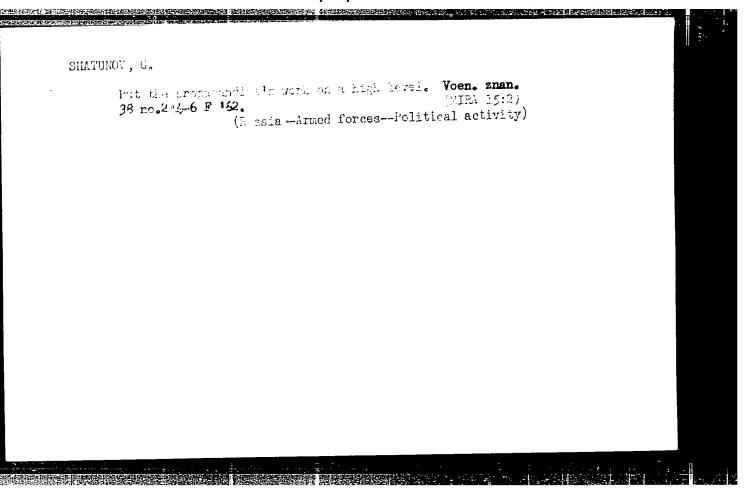
l. Kandidat v chleny Byuro TSentral'nogo Komiteta Vsesoyuznogo Leninskogo Kommumishcheskogo soyuza molodezhi. (Military education)

SHATUNOV, G., general-mayor

We are the stanch followers of Lenin. Komm.Vooruzh.Sil 2
no.18:29-34 S '62. (MIRA 15:8)

1. Chlen prezidiuma TSentral'nogo komiteta Dobrovol'nogo obshchestva sodeystviya armii, aviatsii i flotu.

(Military education)



#### 

SHATUNOV, G., general-mayor

Celebrated patron. Voen. znan. 38 no.10:2-3 0 '62. (MIRA 15:10)
(Communist Youth Lague) (Russie - Havy)

shatunov, g.  $\sqrt{c}$ 

The main direction. Kryl. rod. 14 no.2:7-9 F '63. (MIRA 16:4)

1. Chlen prezidiuma TSentral'nogo komiteta Dobrovol'nogo obshchestva sodeystviya armii, aviatsii i flotu.

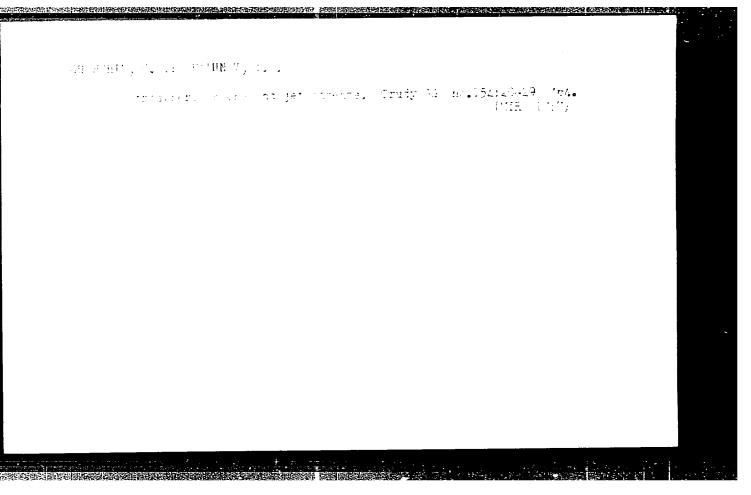
(Aeronautics-Societies, etc.)

Mass participation and skill form the motto of the "Spartekiada."

Kryl.rod. 14 no.9:1-4 S '63. (MEA 16:9)

1. Chlen Prezidiuma TSentral'nogo komiteta Dobrovol'nogo obshchestva sodeystviya armii, aviatsii i flotu.

(Aerial sports)



#### CIA-RDP86-00513R001548720007-7 "APPROVED FOR RELEASE: 08/09/2001

SHATLAN

USSR/MATHEMATICS/Functional analysis SUBJECT

CARD 1/2 PG - 41

AUTHOR

PERIODICAL

SATUNOV M.P.

TITLE

On the resolvent of an elliptic operator. Mat. Sbornik, n. Ser. 37, 459-470 (1955)

reviewed 5/1956

By separation of the variables the operators

Lu + ku 
$$\frac{\text{def}}{\text{ef}} \left[ r_1^2(x_2) + r_2^2(x_2) \right]^{-1} \sum_{i=1}^2 \frac{1}{p_i(x_i)} \frac{\partial}{\partial x_i} \left( p_i(x_i) \frac{\partial u}{\partial x_i} \right) + ku_i$$

infinite;  $x = (x_1, x_2)$ ) decompose into the operators:

$$L_{i} = [p_{i}(x_{i})]^{-1} d/dx_{i}(p_{i}(x_{i})d/dx_{i}) + kr_{i}^{2}(x_{i})$$

$$\Gamma_{i^{u}} = \begin{cases} -\partial u/\partial x_{i} + h u |_{x_{i}=a_{i}} & i = 1,2. \\ \partial u/\partial x_{i} + h_{i}^{!} u |_{x_{i}=b_{i}} \end{cases}$$

The resolvent  $R_{\lambda}$  of the operator L (with the boundary condition  $\Gamma u = 0$ ) is

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Mat. Sbornik, n. Ser. 37, 459-470 (1955) CARD 2/2 PG - 41

represented by means of the resolvents  $R_{-2,i}$  of the operators  $L_i$  (with the boundary conditions  $\Gamma_i u = 0$ ). The following theorem is proved: Let be  $F = (r_1^2 + r_2^2)f \; ; \; r_i(x_i) > 0. \; \text{If } (r_1^2 + r_2^2) \; f \in L^2(\Omega_2), \; F(-,x_i) \in D(L_{s-i}), \; i=1,2$  (D(B) denotes the domain of definition of B), then the function

$$u(-,k) \stackrel{\text{def}}{=} -(2\pi i)^{-1} \int_{C} R_{\lambda,1}R_{-\lambda,2}Fd\lambda$$

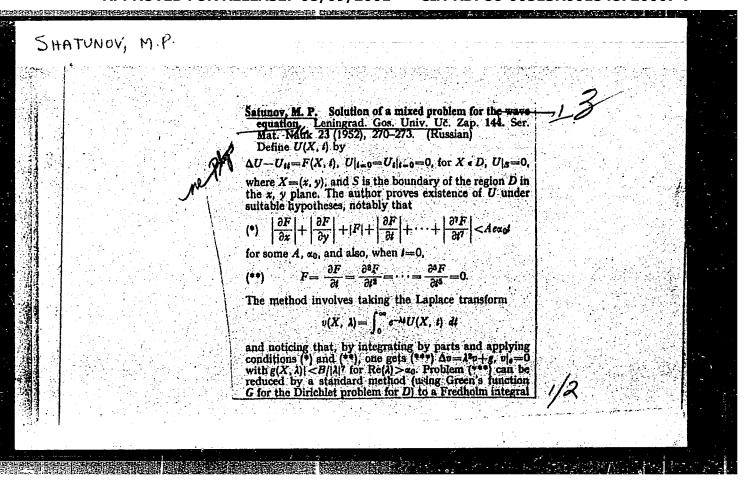
is the solution of the boundary value problem: Lu + ku = f,  $\Gamma$ u = 0 (C is the straight line: arg z =  $\theta > 0$ ). If besides  $r_1, r_2$  are monotonely increasing, then we have

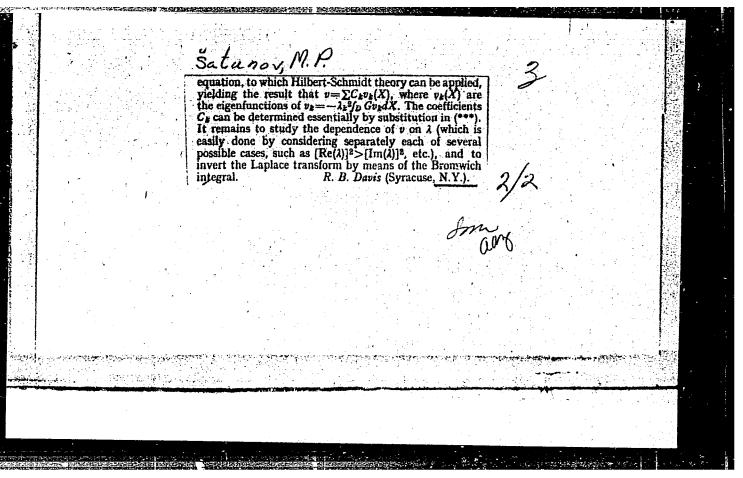
$$R_{k}f = -(2\pi i)^{-1} \int_{-\infty}^{\infty} R_{\lambda,1} R_{-\lambda,2} Fd\lambda \quad (\exists m \ k>0);$$

$$\exists m \ R_{k}f = \pi^{-1} \int_{-\infty}^{\infty} \exists m \ R_{\lambda,1} R_{-\lambda,2} Fd\lambda.$$

INSTITUTION: Kuiby ev

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SOVKIN, V.F., dorsent, kand.tekhn.nauk; SHATUNOV, M.P., dotsent, kand. fiziko-matem.nauk

Calculating temperatures due to grinding. Izv.vys.ucheb. zav.; mashinostr. no.1:131-135 '59. (MIRA 13:3)

 Kuybyshevskiy industrial'nyy institut. (Grinding and polishing)

SHATUNGV, M.P., kand.fiz.-mat.nauk; SOVKIN, V.F., kand.tekhn.nauk

Analytic investigation of the temperature field during grinding based on the general laws of heat conductivity. Izv.vys.ucheb.zav.; mashinostr. no.2:212-223 '62. (MIRA 15:5)

1. Kuybyshevskiy industrial'ny, institut. (Grinding and polishing) (Heat--Conduction)

Shallow, A.r., kami, horke-matematicheskikh nauk, dotsent, SUF 16, 7.F., home, tekhn.nauk, dotsent

Laresti ating contact temperature during grinding basing on natural boundary conditions. Vest.mashincstr. 12 no.5153-27

Je 102. (MIRA 15:5)

(Grinding and polishing)

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denoting limits between his FPV, Fragenty Viktorvier;

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Cair Co. To each to be established at of temperatures and thermal formers are to be established at of temperatures and thermal formers are an end of the established at the control of the established at t

SVISTUNOV, A.M. [deceased]; BEYTEL'MAN, A.N.; IVASHCHENKO, I.Ya.; SHATUNOV, S.F.

Improving certain elements of an open-hearth furnace. Metallurg 8 no.7:17-19 Jl '63. (MIRA 16:8)

VOINCY, S.G.; KOSOY, L.F.; MOROZEMSKIY, A.I.; SAVELLYEV, D.F.; SHALIMOV. A.G.; KALINNIKOV, Ye.S.: SHATUNCY, S.F.; KIREYEV, B.A.; OKHAPKIN, S.I.; DAVYDOVA, L.N.; IZMANOVA. T.A.

CONTROL OF THE EXCENSION OF THE EXCENSIO

Refining a 100-ton open-hearth heat with a liquid synthetic slag in the ladle. Stal' 24 no.7:599-604 Jl '64.

(MIRA 18:1)

MARUSOV, A.Ya., inzhener-podpolkovnik, glavnyy red.; KUDRYAVTSEV, M.K., general-leytenant tekhnicheskikh voysk, otvetstvennyy red.; DEMIN, L.A., inzhener-kontr-admiral, red.; SHCHERBAKOV, A.N., general-mayor, red.; NIKOLAYEV, A.S., polkovnik, red.; KOLOMIYETS, A.D., polkovnik, red.; NAZAROV, P.V., polkovnik, red.; PAROT'KIN, I.V., polkovnik, red.; PUDIKOV, M.P., polkovnik, red.; SISELIN, S.V., polkovnik, red.; BARANOV, M.Kh., inzhener-polkovnik, red.; KOMKOV, A.M., inzhener-polkovnik, red.; SHATUNOV, S.G., inzhener-polkovnik, red.; KOROLEV, V.G., polkovnik, tekhn. red.; TUK'YANOV, B.I., polkovnik, tekhn.red.; ROMANOV, M.K., podpolkovnik, tekhn.red.; IVANOV, V.V., inzhener-podpolkovnik, tekhn.red.; KNYSH, P.N., podpolkovnik tekhnicheskoy sluzhby, tekhn.red.; VASMUT, A.S., kapitan, tekhn.red.; KOSTIN, A.G., tekhn.red.; MAKUKHINA, G.P., tekhn.red.

STATE OF THE PROPERTY OF THE P

[World atlas] Atlas mira. Moskva, Voen.izd-vo M-va obor. SSSR, 1958. 459 p. (MIRA 11:5)

1. Russia (1923- U.S.S.R.) Armiya. General'nyy shtab. Voyennotopograficheskoye upravleniye. 2. Tekhnicheskaya redaktsiya
Voyenno-topograficheskogo upravleniya General'nogo Shtaba (for
Korolev, Luk'yanov, Romanov, Ivanov, Lyubkov, Knysh, Vasmut)
(Atlases)

SHATUNOV, T.

Let's count the seconds. Prom.koop. 13 no.3:7 Mr '59.

(MIRA 12:4)

1. Starshiy inzhener-normirovshchik normativno-issledovatel'skoy stantsii po trudu Kazpromeoveta, Alma-Ata.

(Aktyubinsk-Labor productivity)

MCROZOV, I.1.: CHESTAROV, A.I., inzh., retsenzent; SHATUNOV, V.G., inzh., red.; ZHITROVA, N.A., tekhn. red.

[Organization of train traffic on lenthened haul distances]
Organizatsiia dvizheniia poezdov na udlinennykh tiagovykh
plechakh. Moskva, Transzheldorizdat, 1963. 84 p.

(MIRA 16:10)

(Railroads--Traffic)

ZEMBLINOV, S.V.; STRAKOVSKIY, I.I.; KARLOVSKIY, S.A., inzh.,
retsenzent; SHATUNOV, V.G., inzh., red.; USENKO, L.A.,
tekhn. red.

[Stations and junctions] Stantsii i uzly. Moskva, Transzheldorizdat, 1963. 347 p. (MIRA 17:2)

S/079/61/031/009/009/012

15-8150

1372, 2203, 2405

D215/D306

AUTHORS:

Petrov, K.A., Gavrilova, A.I., Shatunov, V.K., and

Korotkova, V.P.

TITLE:

Diethyleneimides of  $\beta$ -aminoethylphosphinic and

thiophosphinic acids. II

PERIODICAL: Zhurnal obshchey khimii, v. 31, no. 9, 1961,

-3076 - 3081

TEXT: The present work is a continuation of an earlier work, in which the authors showed that diethyleneimidovinylphosphonates and vinylthiophosphonates as well as esters of vinylphosphinic acid vinylthiophosphonates as well as esters of vinylphosphinic acid form addition compounds with mercaptans and alcohols to form corresponding ethyleneimides of alkylphosphinic and alkylthiophosphinic acids. In continuing the investigations, the authors studied the addition of secondary and primary amines to diethyleneimides of vinylphosphinic and vinylthiophosphinic acids. The amines used were diethylamine, ethyleneimine, piperidine, morpholine, dibenzy-

Card 1/3

APPROVED FOR RELEASE: 08/09/2001 CIA-RDP86-00513R001548720007-7" 27508 S/079/61/031/009/009/012 D215/D306

Diethyleneimides of ...

lamine and allylamine; they were found to add to imides of vinyl-phosphinic and vinylthiophosphinic acids to form imides of  $\beta$ -amino-ethylphosphonates and thiophosphonates, according to the following reaction:

$$\begin{array}{c} \text{CH}_2 = \text{CH} - P \left( N \bigvee_{\text{CH}_2}^{\text{CH}_2} \right)_2 + \text{HNR}_2 \longrightarrow R_2 \text{NCH}_2 \text{CH}_2 P \left( N \bigvee_{\text{CH}_2}^{\text{CH}_2} \right)_2. \\ \text{O(S)} \end{array}$$

Diethylamine, piperidine and ethyleneimine readily combine at room temperature over a period of 1.5-2 days or at 40-50°C. for 4-5 hrs. Dibenzylamine and allylamine react in the presence of catalytic quantities of sodium alcoholate. In all cases it is advisable to use equimolecular quantities without a solvent. Addition of amines to the imides of the acids is more difficult than in the case of the addition of amines to neutral esters of the acids. The addition products of piperidine, morpholine and diethylamine with the imides of the acids were purified by vacuum distillation (10-4 mm); the products of the other amines decomposed on distilling. All di-

Card 2/3

27508

Diethyleneimides of ...

S/079/61/031/009/009/012 D215/D306

ethylene-imides of aminophosphonates and aminothiophosphonates were viscous, colorless liquids, soluble in benzene, chloroform, ether and alcohol and are stable at temperatures below 0°C. Prolonged storing at room temperature results in gradual polymerization which is due to the opening of the ethyleneimide rings and results in the production of linear polymers either without a phosphorus residue or with the phosphorus residue binding the main chains of the macromolecule. The compounds which were prepared and their properties are summarized in tabulated form. Preparation of compounds and 9 was conducted at room temperature and of compounds 8, 10, bles and 3 Soviet-bloc references.

SUBMITTED: September 5, 1960

Card 3/3

27509

15-8150

S/079/61/031/009/010/012

D215/D306

AUTHORS:

Petrov, K.A., Gavrilova, A.I., Shatunov, V.K., and

Korotkova, V.P.

TITLE:

Diethyleneimides of alkyl- and alkenylthiophosphinic

and phosphinic acids. I

PERIODICAL: Zhurnal obshchey khimii, v. 31, no. 9, 1961,

3081 - 3085

TEXT: The authors studied the properties of diethyleneimides of alkyl- and alkenylthiophosphinic and allylphosphinic acids, and investigated the addition of mercaptans and alcohols to diethyleneimides of vinylphosphinic and vinylthiophosphinic acids. Their aim was to prepare imidophosphonates and thiophosphonates containing ether and thioether groups in a radical bonded with phosphorus through carbon. Diethyleneimides of alkyl- and alkenylthiophosphinic and allylphosphinic acids were prepared by reacting the corresponding acid chlorides with ethyleneimine in dry benzene or ether

Card 1/3

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27509 S/079/61/031/009/010/012

D215/D306

Diethyleneimides of alkyl- ...

in the presence of a tertiary base (HCl acceptor) at 5-10°C according to the following reaction:

$$\begin{array}{c|c}
R - PC1_{2} \\
S(0) + 2HN
\end{array}$$

$$\begin{array}{c|c}
CH_{2} \\
-R_{3}N & HC1
\end{array}$$

$$\begin{array}{c}
R - P \\
S(0)
\end{array}$$

$$\begin{array}{c}
CH_{2} \\
CH_{2}
\end{array}$$

The products were colorless liquids, readily soluble in water and organic solvents; some of them crystallized on prolonged standing. Almost all the compounds distilled in vacuum, the one exception being the diethyleneimide of  $\beta$ -chloroethylthiophosphinic acid which polymerizes at  $100-102^{\circ}\mathrm{C}$  and  $10^{-4}\mathrm{mm}$  pressure probably due to HCl splitting off which initiates spontaneous nolymerization. The properties and yields of some of the prepared phosphinates and thiophosphinates are given in tabulated form. Diethyleneimides of vinylphosphinic and thiophosphinic acids form addition products with mercaptans and alcohols. With mercaptans the reaction occurs at  $60^{\circ}\mathrm{C}$  and is complete in 14-15 hrs. or less if catalytic quantities of sodium alcoholate is present. Ethylmercaptan adds more

Card 2/3

27509 S/079/61/031/009/010/012 D215/D306

Diethyleneimides of alkyl- ...

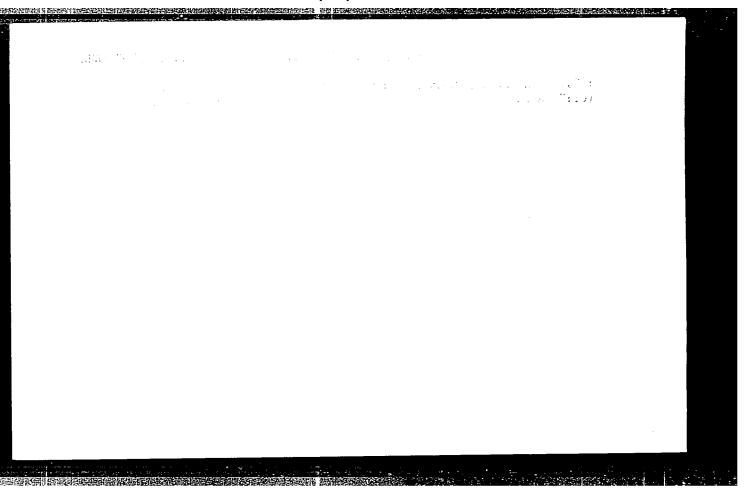
easily than butyl mercaptan and in general the reaction proceeds as follows:

$$CH_{2} = CHP \left(N \left(\frac{CH_{2}}{CH_{2}}\right) \xrightarrow{RSH} RSCH_{2}CH_{2}P \left(N \left(\frac{CH_{2}}{CH_{2}}\right)\right).$$

The yields are in the region of 50-60 %. Alcohols unlike mercaptans add less readily and it was possible to obtain only small yields of ethyl and butyl alcohol addition products, only after prolonged heating in the presence of alcoholates. Better yields were obtained by reacting alcoholates with diethyleneimides of  $\beta$ -chloroethylphosphinic acid. There are 2 tables and 3 references: 2 Soviet-bloc and 1 non-Soviet-bloc. The reference to the English-language publication reads as follows: U.S. Pat. 2,654,738, 1953; U.S. Pat 2,672, 459, 1952.

SUBMITTED: September 5, 1961

Card 3/3



L 6857-65 EWT(d)/EEC(k)-2/EEC; Pg-4/Pk-4/Pl-4/Po-4/Pq-4 SSD/ASD(a)-5/
AFWL/ESD(t)/RAEM(t)
ACCESSION NR: AR4044266 S/0272/64/000/066/0064/0064 74

SOURCE: Ref. zh. Metrologiya i izmeritel'naya tekhnika. Otdel'ny\*y vy\*pusk, Abs. 6.32.443

AUTHOR: Vyaselev, M. R.; Shatunov, V. S.

TITLE: The dipole moment meter IDM-2

CITED SOURCE: Nauchn. tr. vuzov Povolzh'ya, vy\*p. 1, 1963, 250-252

TOPIC TAGS: dipole moment, molecular dipole moment, capacitance, capacitor, dipole moment meter, zero beat, measuring instrument/IDM-2 dipole moment meter, IDM-2 meter

TRANSLATION: Determination of the dipole moments of molecules involves measurement of the permittivity of liquids and the solutions. Such measurements can be made with the help of the instrument IDM-2. The investigated substance is poured between the plates of a special capacitor of the cell, and from the change of its capacitance is determined the permittivity of the substance. For Card 1/2

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L 6857-65 ACCESSION NR: AR4044266

determination of the change of capacitance of the cell the "zero beat" method is used. A sweep generator generates a sinusoidal voltage, which is fed to a mixer. A measuring capacitor of variable capacitance and a cell are connected in parallel in the generator circuit. The measuring capacitor has a reductor with large delay and a scale graduated in conventional units of capacitance. A fixed-frequency generator feeds the stabilized sinusoidal voltage to the mixer. The instrument is tuned to zero beat (the difference of frequencies is equal to zero) by alternately filling the cell with known and control solutions. The permittivity is determined from the difference of the readings of the measuring capacitor. Two illustrations.

SUB CODE: EM ENCL: 00

Card 2/2

5/0032/64/030/004/0500/0501

ACCESSION NR: APLO33621

AUTHORS: Nigmatullin, R. Sh.; Vyaselev, M. R.; Shatunov, V. S.

TITLE: A device for dipole moment measurements IDM-2

SOURCE: Zavodskaya laboratoriya, v. 30, no. 4, 1964, 500-501

TOPIC TAGS: dipole moment, dielectric constant, measuring device IDM 2, beat frequency method, dimethylformamide, chlorobenzene, phenylhydrazone acetaldehyde

ABSTRACT: An IDM-2 device for determining the dipole moment of molecules is described. The dielectric constant of a dilute solution in a nonpolar solvent is measured in a specially constructed capacitor which consists of two concentric glass cylinders between which the investigated solution is poured. The capacitor plates are ordinary foil wrapped around the outside of the larger cylinder and the inside of the smaller. Hence the special capacitor  $C_{\rm S}$  is a series combination of two capacitors: one formed by the glass cylinders  $C_{\rm C}$  and the other by the filled gap  $C_{\rm D} = \{C_{\rm O}, \text{ where } C_{\rm O} \text{ is the capacitance of the empty gap. The special capacitor is placed in parallel with a precision variable capacitor. The resultant capaci-$ 

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ACCESSION NR: AP4033621

tance controls the output frequency  $f_1$  of a signal generator. The frequency  $f_1$  is mixed with a constant frequency  $f_0$  stabilized by a quartz resonator, which produces a heat frequency  $f_1 - f_0$ . The variable capacitor is adjusted until the heat frequency is zero. If the capacitance of the special capacitor  $C_{\rm sl}$  with a known control solution and  $C_{\rm sx}$  with the investigated solution, the difference of the two is the same as the difference  $\Delta C$  required in the variable capacitor to produce zero beat frequency for the two cases. Then the dielectric constant of the investigated solution is given by

$$\varepsilon_x = \frac{C_c (C_{S1} + \Delta C)}{C_0 [C_c - (C_{S1} + \Delta C)]} \dots$$

The temperature of the special capacitor is thermostatically controlled. Thus the temperature dependence of the dielectric constant can be determined, an example of which is given for dimethylformamide. The dielectric constant can be measured to 0.05% accuracy for  $\xi=1-3$  and 0.5% for  $\xi$  up to 100. The dipole moment can be computed by the Debye formula for dilute solutions or by the Onsager formula for pure liquids. As an example, the dipole moments of chlorobenzene and phenylhydra-

Card 2/3

ACCESSION NR: AP4033621

zone acetaldehyde (measured in benzene solutions at 25C) were found to be 1.59 and 2.52 respectively. Orig. art. has: 2 equations, 1 diagram, and 2 tables.

ASSOCIATION: Kazanskiy aviatsionny\*y institut (Kazan Aviation Institute)

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Card 3/3

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## "APPROVED FOR RELEASE: 08/09/2001

### CIA-RDP86-00513R001548720007-7

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S/081/61/000/003/006/019 A166/A129

AUTHORS:

Yefimova, A. K., Shatunova, A. M., Vol'f, M. B.

TITLE:

Selecting hydrogen chloride and hydrogen sulfide corrosion inhibitors

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 3, 1961, 300, abstract 31176. (Tr. Bashkirsk. n.-i. in-ta po pererabotke nefti, 1960, no. 3, 181 -

203)

A study of the effects of various organic corrosion inhibitors on the TEXT: corrosion of grade CT 3 (St 3) steel samples at a temperature of  $80^{\circ}$ C in hydrogen chloride and hydrogen sulfide media showed that nitrogenous bases from heavy petroleum products of coal-tar pitch distillates, high-molecular amines (C15-C18) and hydroxyethyl heptadicenylglyoxalidine are effective corrosion inhibitors in weak HCl and HoS solutions at 80°C.

Summary by M. Platkov

[Abstracter's note: Complete translation]

Card 1/1

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s/081/62/000/001/037/067 B102/B101

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AUTHORS: Yefimova, A. K., Vol'f, M. B., Shatunova, W. M.

TITLE: Nitric bases of petroleum and their use as corrosion

inhibitors

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 1, 1962, 313, abstract

11243 (Sb. "Khimiya seraorgan. soyedineniy,

soderzhashchikhsya v neftyakh i nefteproduktakh. v. 4". K.,

Gostoptekhizdat, 1961, 265-268)

TEXT: Use of nitric bases extracted from vacuum gas oil as anticorrosive agents has shown that, when they are added in amounts of 0.1%, the corrosion rate of mild steel in the gasoline-condensation water of the ABT(AVT) is reduced by 80-90%. [Abstracter's note: AVT stands for atmospheric-vacuum pipe still.] If both nitric bases and ammonia are added, a 90% corrosion protection can be reached when each of the additives amounts to 0.005%. [Abstracter's note: Complete translation.]

Card 1/1

S/081/62/000/022/035/088 B158/B101

AUTHORS: Yefimova, A. K., Shatunova, A. M., Sapozhnikova, Ye. A.

TITLE: Experience in industrial tests for corrosion inhibitors

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 22, 1962, 307, abstract

22I2OO (Novosti neft. i gaz. tekhn. Neftepererabotka i nefte-

knimiya, no. 2, 1962, 46 - 47)

TEXT: A number of corrosion inhibitors (CI) have been selected for protecting equipment in the petroleum industry from corrosion. These CI reduce the rate of corrosion of ferrous metals by 70 - 95 % and of tin brass by 50-60%. At present sulfosodium salts of shale tar and nitrogenous petroleum compounds are the most available CI. Introducing ammonia up to pH 7 - 9 reduces the CI consumption to 1/10 - 1/20. Data are given on the rate of corrosion of various metals in the condensation system of an atmospheric-vacuum pipe still during the processing of Tuymazino oil and on the efficiency achieved as a result of CI introduction. A particularly sharp fall in corrosion was found for A1, for which 99 % protection was obtained. Abstracter's note: Complete translation.

Card 1/1

YEFIMOVA, A.K.; SHATUNOVA, A.M.; SAFOLHNIKOVA, Ye.A.

Corrosion innibitors for protecting the condensation-cooling apparatus of atmospheric and vacuum distillation units. Trudy Bash NINP no.5: 165-175 162. (MIRA 17:10)

SHATUNOVA, G.N.

New technology for growing fodder yeasts. Gidroliz. i lesokhim. prom. 14 no.8:21-23 '61.

1. Tavdinskiy gidroliznyy zavod.

SHATUNOVA, N.F.

TREATMENT OF THE PROPERTY OF T

Intracellular localization of gamma-aminobutyric and glutamic acids. Biokhimiia 29 no.4:647-652 Jl-Ag 164.

(MIRA 18:6)

l. Laboratoriya khimii belka Leningradskogo ordena Lenina gosudarstvennogo universiteta imeni Zhdanova.

The effect of semicarb zide poisoning on the content of 2°-aminobutyric acid in brain tissues. Nerv. sist (Leningrad) 2 no.3:12-16 '52. (KIRA 17:7)

1. Laboratoriya khimii belka Fizirlevioleskigo instituta imeni Ukhtomskogo Leningraiskogo posudaratvennogo universiteta.

SHATUNOVA, N.F.

Intracellular localization of glutamic decarboxylase. Vop.med. khim. 10 no.3:322-323 My-Je '64. (MIRA 18:2)

1. Laboratoriya khimii belka Leningradskogo gosudarstvennogo universiteta.